Compensated magnetism by design in double perovskite oxides

Victor Pardo^{1,2,*} and Warren E. Pickett¹

¹Department of Physics, University of California, Davis, California 95616, USA

²Departamento de Física Aplicada, Facultad de Física, Universidad de Santiago de Compostela, Campus Sur s/n,

E-15782 Santiago de Compostela, Spain

(Received 25 May 2009; published 25 August 2009)

Taking into account Goodenough's superexchange rules, including both full structural relaxation and spinorbit coupling, and checking strong correlation effects, we look for compensated half metals within the class of oxide double perovskites materials. Identifying likely half-metallic (or half-semimetallic) antiferromagnets, the full complications including orbital magnetism are included in order to arrive at realistic predictions of designed magnetic compounds with (near) vanishing net moment. After sorting through several candidates that have not been considered previously, two materials, K₂MnRhO₆ and La₂CrWO₆, remain as viable candidates. An important factor is obtaining compounds either with very small induced orbital moment (helped by closed subshells) or with an orbital moment that compensates the spin-orbit driven degradation of half-metallic character. While thermodynamic stability of these materials cannot be ensured, the development of layer-bylayer oxide deposition techniques does not require that materials be thermodynamically stable to be synthesized.

DOI: 10.1103/PhysRevB.80.054415 PACS number(s): 75.30.Et, 72.25.Ba, 72.25.Dc, 71.20.-b

I. INTRODUCTION

Double perovskites have drawn considerable attention in the field of spintronics, particularly since the discovery¹ of colossal magnetoresistance in Sr₂FeMoO₆ that is believed to be a half-metallic (HM) ferromagnet (FM). The HM property is highly desired in spintronics applications, where spin currents are utilized as well as charge currents. The double perovskite family of oxides has been one of the most popular classes within which to look for HMFMs.²

Recently, with wider application of deposition techniques leading to better materials and improved understanding, layer-by-layer deposition of perovskite oxides has led to unusual and potentially useful physical properties.³ Most such depositions of perovskites have used the (001) growth direction, resulting in a wide variety of oxide heterostructures with tetragonal symmetry. Double perovskite materials can be considered as single unit cell, multilayered perovskite structures that have been grown along the (111) direction, and such growth is a promising direction for synthesizing previously unexplored members of this class.

Although HMFM character will provide previously unavailable (and highly sought after) magnetoelectronic properties, there is a subclass that is more exotic. When half metallicity occurs and in addition the moments compensate, a "compensated half metal" (historically called a halfmetallic antiferromagnet⁴) arises. This state has magnetic order but vanishing macroscopic spin magnetization, it could support a peculiar spinless type of superconductivity,⁵ and one can imagine numerous possible applications for a half metal that is relatively impervious to external magnetic fields. However, this spin-only picture is an idealization; spin-orbit coupling (SOC) couples the two spin directions and thereby destroys the precise spin compensation, and in addition generates spin-induced orbital moments. Thus a fully compensated half metal with vanishing macroscopic magnetization is an idealization and the focus must be on finding realizations that are as close as possible to full compensation.

Interest has been rekindled recently in the effects of spinorbit coupling in oxides because, unlike the degradation of half metallicity that is commonly discussed and expected, it can also lead to more exotic effects in the spin and orbital magnetism, especially in d^1 or d^5 ions in t_{2g} subshells where large orbital moments may arise.⁶ A dramatic possibility is the promotion of compensating spin and orbital moments. In Ba_2NaOsO_6 , for example, calculations⁷ indicate that the d^1 spin moment of Os is compensated by the t_{2g} L=1 orbital moment induced by SOC, and it is the partial orbital quenching by the environment that destroys the compensation of the moment. In Sr₂CrOsO₆, neglect of SOC leads to a semimetallic compensated half metal, which is destroyed by the large SOC in this system. 8 In isovalent Sr₂CrRuO₆, where weaker SOC arises on Ru compared with Os, the reduced spin and induced orbital moments conspire to nearly cancel, giving a real possibility for an effectively compensated half metal. In correlated insulators, strong SOC can lead to much more complex magnetic coupling, thereby complicating the resulting magnetic order⁹ and reducing the likelihood of achieving a compensated half metal.

Half-metallic antiferromagnetic (HMAF) materials have not been obtained experimentally yet. Several theoretical efforts have been carried out in predicting half-metallic antiferromagnetism but experimentally, this state remains elusive. HMAFs have been predicted for various double perovskites, ^{10–13} tetrahedrally coordinated compounds, ¹⁴ and for Heusler-structured materials. ^{15–17} Design of materials with desired properties is a long-standing hope that is gaining momentum ¹⁸ and the considerable experience that has been accumulated in oxide double perovskites provides guidelines to focus design efforts.

In this paper we build on previous studies and extend the earlier work by considering all of factors: ion size, structural relaxation, and SOC. Our general strategy is: (i) we focus on the double perovskite structure, with the chemical formula

A₂MM'O₆, where M and M' are transition-metal ions lying on a rocksalt-type sublattice. An ordered double perovskite is more likely to occur if M and M' differ in ionic radius and differ in formal charge. (ii) We use ab initio calculations to relax the structure, including volume optimization, c/a optimization, and internal coordinate relaxation, thereby obtaining structures that are dynamically stable, and energetically metastable if not stable. (iii) We include SOC, which mixes the spin moment with an orbital moment and also analyze the effects of correlation corrections by using the LSDA +U method, 19,20 to check the dependence of the magnetic and electronic structure properties on U (the on-site Coulomb repulsion). The objective is to determine to what degree the spin compensation is maintained and to understand the interplay of structural changes, SOC, and effects of electronic correlation.

We use the Goodenough-Kanamori-Anderson (GKA) rules for superexchange²¹ for the 180° metal-oxygen-metal bond occurring in perovskites to guide our choices. According to these authors, and the expected moments, only a few d^n - d^m electronic configurations that would lead to a spincompensated situation are likely. In addition, we aim to maximize the likelihood of obtaining an ordered perovskite by mixing a 3d row element with another metal cation from the 4d or 5d row so that the size mismatch will make chemical ordering more likely. According to GKA, the possible antiferromagnetically coupled electronic configurations are d^2 - d^2 , d^2 - d^4 (LS), d^3 - d^3 , d^8 - d^8 and other combinations that include in all cases high-spin (HS) cations: d^4-d^4 , d^5-d^5 , d^4 - d^6 , d^6 - d^6 , and d^7 - d^7 . These latter options have not been pursued because it is difficult to stabilize HS states with those large moments for the 4d and 5d elements, which usually take on a higher valence state than the 3d element, and hence the relatively strong crystal field will favor low-spin states. Particularly interesting is the d^3 - d^3 configuration when SOC is taken into account. Since it deals with a completely filled t_{2g} band, the effective angular momentum will be very small, although there is also the possibility of insulating states. We will discuss this issue later in the paper.

II. COMPUTATIONAL DETAILS

Electronic-structure calculations were performed within density-functional theory²² using WIEN2K,²³ which utilizes an augmented plane wave plus local orbitals (APW+lo) (Ref. 24) method to solve the Kohn-Sham equations. This method uses an all-electron full-potential scheme that makes no shape approximation to the potential or the electron density. The exchange-correlation potential used was the Perdew and Wang version of the local-density approximation²⁵ and strong correlation effects were introduced by means of the LSDA+U scheme^{20,26} including an on-site U (Coulomb repulsion) and J (Hund's rule exchange) for the metal d states. We have used option nldau=1 in WIEN2K, i.e., the so-called "fully localized limit." Structural optimizations (unless otherwise stated) were performed using the Perdew-Burke-Ernzerhof version of the generalized approximation.²⁷ Spin-orbit coupling was introduced using the scalar relativistic approximation. All the calculations

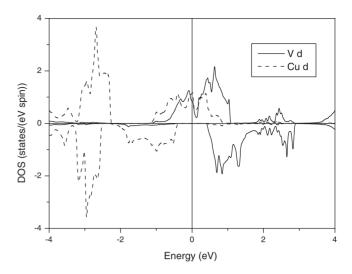


FIG. 1. Density-of-states plots of the tetragonal double perovskite compound $\rm La_2VCuO_6$ in a HMAF state within an LSDA+U scheme with U=3.5 eV for both V and Cu. Upper (lower) panel shows up (down) spin channel. The spin moment is precisely compensated.

were converged with respect to all the parameters used, to the precision necessary to support our calculations (converged forces and total-energy differences), up to $R_{mt}K_{max}$ = 7 and an $8\times8\times8$ k mesh.

III. REVISITING PREVIOUSLY FOUND MATERIALS

A. La₂VCuO₆: d^1 - d^9

It had been reported previously that this material is a HMAF (Refs. 10 and 28) but previous studies did not consider the structural relaxation that we have now carried out, and that is expected to be important in this strongly Jahn-Teller active system. At the LSDA+U level, this compound converges to a HMAF, which remains when one relaxes the structure. We have done a relaxation within a tetragonal symmetry (but the ratio $c/a \sim 1.00$, to 1% accuracy, as expected), obtaining a=3.64 Å. The internal structure relaxes to accommodate the Jahn-Teller distortion of the Cu²⁺ and V⁴⁺ cations. For this reason, it is interesting to analyze the lattice distortions including strong correlation effects, that allows the unpaired electron or hole to be largely localized in one particular orbital, leading to a Jahn-Teller-type distortion. Depending on the value of U chosen (the same U was chosen for both cations in our calculations), this distortion varies, being larger for a larger U. For relatively small values of U (effective U of only 4 eV) there is a large Jahn-Teller distortion, the oxygen octahedra around the Cu²⁺ cation become elongated by about 4% to accommodate the d^9 cation and around V⁴⁺, the octahedron is shortened along one axis by about 4% to accommodate a d^1 cation. The system becomes a Mott insulator for U=4.8 eV and J=0.7 eV. The electronic structure is presented in Fig. 1 for U=3.5 eV, where the system is still half metallic.

Figure 1 shows that half metallicity occurs with the Fermi level falling within a 2 eV wide band of Cu e_g and V t_{2g} character, even when degeneracy is broken by the Jahn-

Teller distortion. The bottom of the band (a two-band complex) is strongly $\operatorname{Cu} d$ in character while the $\operatorname{V} d$ character dominates the upper part of the bands.

We expect a small influence of SOC because only 3d elements are involved. The full moment compensation that occurs within LSDA+U will be kept when including SOC. This d^9 - d^1 configuration has small magnetic coupling according to GKA rules (AF according to our calculations), and is spin compensated according to our results. The orbital angular momentum of the Cu atoms is $0.1~\mu_B$ but is negligible for V and the electronic structure is not affected by the introduction of SOC. Spin compensation remains unaltered when SOC is introduced, thus the total moment of the system is approximately $0.1~\mu_B/\mathrm{f.u.}$, quite close to the desired result.

B. $Sr_2NiOsO_6:d^8-d^2$

This material has been synthesized²⁹ recently and also has been studied using ab initio methods.³⁰ Experimental results were interpreted in terms of a competition between FM and AF interactions, caused by the d^8 - d^2 (Ni²⁺-Os⁶⁺) electronic structure. The closed subshell (d^8) and open subshell (d^2) should favor half metallicity. According to GKA rules, the e_{ρ}^{2} -O- e_{ρ}^{0} FM superexchange should be the strongest in this system. However, the magnetic-susceptibility data shows both a positive Θ_{CW} (showing that FM is dominant at high temperature) and an AF downturn below 50 K, indicating that AF correlations exist at low temperature. Our ab initio calculations show the FM and AF configurations are similar in total energy but FM is distinctly the ground state by 20 meV/metal atom (using the experimental structure). Also, the spin compensation that results within LSDA+U is degraded seriously by SOC, with the large orbital angular momentum leading to a total moment of approximately 1 $\mu_B/f.u.$ This value arises from orbital moments of 0.5 μ_B for Os and 0.2 μ_B for Ni, plus the 0.2 μ_B from the total spin moment of the system arising from SOC-induced spin mixing. Hence, the condition of a small magnetic moment is not obtained for this compound. The d^2 - d^8 electronic structure has the drawback that, when an antialigned spin configuration is set, the orbital angular momenta align and add up rather than (partially) canceling (Hund's third rule for the atoms).

We have performed a structural analysis similar to the one described for La_2VCuO_6 with the same preconditions (in this case strong correlation effects were not introduced in the calculations). Comparing with experimental structural data, ²⁹ our calculated volume is quite good (1% larger than experiment) and the c/a ratio is 1% smaller than the experimental value (Table I).

For large values of U this compound becomes a Mott insulator. Half metallicity occurs only for U smaller than 3.5 eV for Os (with J=0.7 eV) and values of U not larger than 7 eV for Ni. Realistic values of U and U would be close to the metal-insulator transition. No experimental information on its conduction properties is available so far. For smaller values of U, we obtain conduction coming from the Os majority-spin channel (the unfilled subshell), as can be seen in Fig. 2.

TABLE I. Calculated versus experimental structural parameters and selected bond lengths for Sr_2NiOsO_6 showing agreement to within 1% accuracy in the lattice constants.

Parameter	Experimental ^a	This work
a	5.49	5.53
c	7.99	7.94
Ni-O distance (×4)	2.02	1.99
Ni-O distance (\times 2)	2.04	2.03
Os-O distance $(\times 4)$	1.91	1.92
Os-O distance $(\times 2)$	1.96	1.94

^aReference 29.

IV. AB INITIO DESIGNED COMPOUNDS

Two materials with the desired properties have been clearly identified. Both possess a d^3 - d^3 electronic configuration so a high-spin magnetic-band insulator situation is a possibility. This particular electronic configuration has several advantages, that we will discuss, over other possible electronic configurations that will promote a spin-compensated state and they have AF coupling by the GKA rules so they become plausible candidates for being spin-compensated half metals or half semimetals. In addition, we have analyzed a compound with a d^2 - d^2 configuration, and we discuss distinctions.

A. $Sr_2VReO_6:d^2-d^2$

This compound is calculated to have a d^2 - d^2 electronic configuration. In agreement with GKA rules, the AF solution is favorable with respect to the FM solution [by about 50 meV/metal using the LDA+U approach, with U(V)=7 eV and U(Re)=3 eV, being J=0.7 eV for both cations]. The unoccupied t_{2g} shells lead to a distortion from cubic symmetry. Relaxing the structure within tetragonal symmetry, the

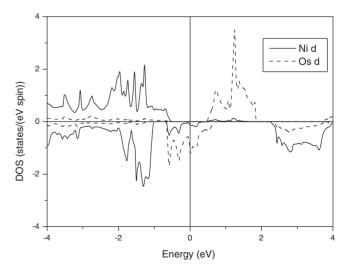


FIG. 2. Density-of-states plots of the tetragonal double perovskite Sr_2NiOsO_6 in a HMAF state within an LSDA+U scheme with U=5 eV for Ni and 3 eV for Os, including spin-orbit coupling. Upper (lower) panel shows up (down) spin channel.

TABLE II. Calculated structural parameters for Sr_2VReO_6 in the space group I4/mmm (no. 139). a=5.52 Å $(\sqrt{2}\times3.90$ Å) and c=7.96 Å $(2\times3.98$ Å).

Atom	Wyckoff position	Atomic parameter
Sr	4 <i>d</i>	(0.5,0,0.25)
V	2a	(0,0,0)
Re	2b	(0,0,0.5)
O1	8h	(0.2521, 0.2521, 0.5)
O2	4e	(0,0,0.2494)

resulting c/a ratio is 1.02, together with a sizable elongation of the oxygen cages to accommodate the d^2 electronic state (3% difference in the V-O distance and a smaller 1% difference for the Re-O distance). A summary of the structural parameters obtained is given in Table II.

We have studied the electronic structure of this yet-unreported compound with both LSDA+U and SOC. Due to slight band overlap, there is substantial but not total spin-moment compensation, leading to a total spin moment of less than 0.1 μ_B . However, because of the unfilled t_{2g} shells in both cations, large orbital angular moments develop, particularly large for Re (0.4 μ_B). The result is a total magnetic moment of approximately 0.5 μ_B for U(V)=7 eV and U(Re)=3 eV (J=0.7 eV for both cations). While there is no true gap in the V majority-spin channel, there is a very large fractional spin polarization in the density of states at the Fermi level, as can be seen in Fig. 3.

B. $K_2MnRhO_6:d^3-d^3$

As we have just seen in the case of Sr_2VReO_6 , having unfilled shells encourages orbital angular moments. In that respect, the following two examples are materials which have a d^3 - d^3 electronic configuration so the filled t_{2g} shells

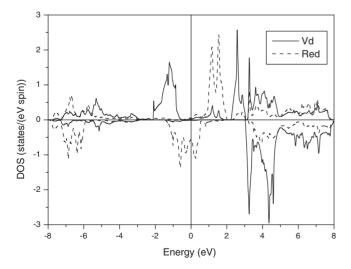


FIG. 3. Density-of-states plots of the tetragonal double perovskite Sr_2VReO_6 in a HMAF state within an LSDA+U scheme with U=5 eV for V and 3 eV for Re, including spin-orbit coupling. Upper (lower) panel shows up (down) spin channel.

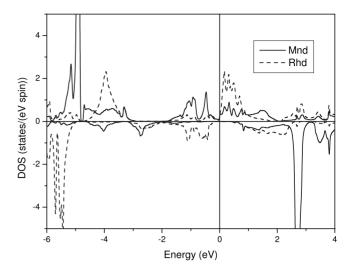


FIG. 4. Density-of-states plots of the tetragonal double perovskite K_2MnRhO_6 in a HMAF state within an LSDA+U scheme with U=5 eV for Mn and 3 eV for Rh, including spin-orbit coupling. Upper (lower) panel shows up (down) spin channel.

will give very small orbital moments to obscure the spin compensation.

We calculate that this K₂MnRhO₆ compound has the configuration Mn⁴⁺:d³-Rh⁶⁺:d³, which couples AF. The totalenergy difference between a FM and an AF configuration is about 20 meV/metal for U=6 eV for Mn and 4 eV for Rh. This result is in agreement with GKA predictions of a weak AF coupling for a d^3 - d^3 superexchange situation, with transition temperatures typically not exceeding 100 K. The Rh moment is strongly delocalized on the oxygen neighbors (up to 0.3 μ_B per oxygen and only 1 μ_B in the Rh muffin-tin sphere) but all in all, this compound is near HMAF with Fermi level lying very near band edges. The lattice parameter a was optimized and atomic positions were relaxed within the $Fm\overline{3}m$ space group (no. 225), and a=3.94 Å was obtained. The d^3 - d^3 configuration of ions with cubic symmetry leaves a cubic structure, leaving only a small relaxation of the oxygen cages around the cations (Rh-O distance of 2.00 Å and Mn-O distance of 1.94 Å).

As mentioned, very small values of the orbital angular momentum arise when SOC is introduced: 0.05 μ_B on Rh and 0.01 μ_B on Mn. Adding spin and orbital angular momenta, the net moment is only 0.02 μ_B and it remains very close to compensated half metal for values of U of 6 eV for Mn and 4 eV for Rh. Even though Fig. 4 looks similar to a zero-gap situation, on the verge of a metal-insulator transition, bands cross the Fermi level even for larger values of U, and (half) semimetal character is maintained.

C. La₂CrWO₆: d^3 - d^3

This is another example of a d^3 - d^3 AF coupling situation and again, it is a not-yet-synthesized material, completely *ab initio* designed. Our calculations yield a very large stability of about 250 meV/metal of the AF solution with respect to the FM one for values of U of 6 eV for Cr and 4 eV for W. Structural relaxation was carried out allowing tetragonal dis-

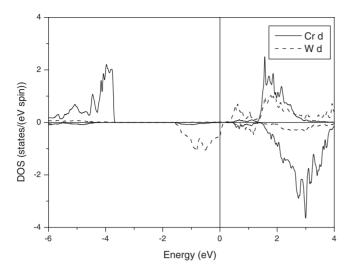


FIG. 5. Density-of-states plots of the tetragonal double perovskite La_2CrWO_6 in a HMAF state within an LDA+U scheme with U=5 eV for Cr and 3.5 eV for W, including spin-orbit coupling. Upper (lower) panel shows up (down) spin channel.

tortion. No tetragonal distortion was obtained and a = 4.00 Å. The difference in ionic radii leads to a Cr-O distance of 1.97 Å and W-O distance of 2.03 Å.

Evolution with U and SOC indicates that the latter does not seriously degrade the moment compensation. For large values of U, where the material is still a metal, the total moment stays below $0.2~\mu_B/f.u.$ W has an orbital moment of $0.15~\mu_B$ and a much smaller one, and opposite in an AF configuration, arises from Cr. Even for large values of U (7 eV for Cr and 4 eV for W), the system remains metallic (half metallic) and the total magnetic moment, including spin and orbital components, does not exceed $0.10~\mu_B/f.u.$ Metallicity comes from a broad W d band that crosses the Fermi level, as can be seen in Fig. 5. An electron pocket is formed in an itinerant W d band and a corresponding La-character hole pocket (not shown) appears.

V. UNPROMISING COMPOUNDS

On our way to detecting these spin-compensated half metals, we studied other seemingly promising double perovskites. We will discuss these only briefly since they do not show the desired behavior.

A. $d^{1}-d^{9}$

Even though superexchange rules predict this coupling to be negligible, we have previously studied a system that could be promising. Hence, as an analogy with the Jahn-Teller compound $\rm La_2VCuO_6$, we tried changing each of the 3d elements by a corresponding 4d element, without changing the electron count, i.e., $\rm La_2CuNbO_6$ and $\rm La_2VAgO_6$ which are less likely to be Mott insulators. In these cases, Nb and Ag, respectively, are nonmagnetic and the spin compensation is lost.

D d2 d4

This electronic configuration could lead to a spincompensated state, if the d^4 cation is in a low-spin state. We have tried Ba_2FeCrO_6 (also with Sr and Ca with similar results), which could assume a d^2 - d^4 electronic configuration. It is indeed the case but Fe is in a HS state (not allowing for spin compensation to happen). A similar problem appears with the compound K_2FeRuO_6 , where also Fe^{4+} is also in a HS state. This electronic configuration has been studied in the past 11,12 for several perovskites with general formula $AA'BB'O_6$.

C. $d^2 - d^8$

We have found several cases where always the FM exchange mediated by the e_g electrons is more intense than the AF coupling between the t_{2g} electrons. Starting from Sr_2NiOsO_6 compound discussed above and substituting Os by Ru to form Sr_2NiRuO_6 , we find that FM coupling remains. This is the standard prediction of GKA rules. Apart from these, we have also identified that FM coupling is more favorable in La_2CrNiO_6 (and also with W and Mo on the Cr site). This material is calculated to be an interesting FM insulator (of which there are relatively few examples), where instead of promoting the possible Jahn-Teller-distorted Ni^{3+} cation, it leads to the formation of a $Cr^{2+}-Ni^{4+}$ configuration. This charge-ordered state is thought to occur to prevent the system from developing a Jahn-Teller distortion when it is close to the itinerant electron limit.³¹

Disordered $LaCr_{0.5}Ni_{0.5}O_3$ is known experimentally to be insulating 32 but close to a metal-insulator transition (the Ni end of the series $LaNiO_3$ is a metal). Our calculations show that the electronic structure of the charge-ordered material explains the insulating state found experimentally, together with the prediction of FM coupling between the d^2 and d^8 cations, making the system potentially interesting as another FM insulator. For these calculations, we have taken the lattice parameters from the experiment 33 with the disordered system and relaxed only the internal coordinates. The different ionic radii lead to different metal-oxygen bond lengths (1.90 Å for Cr and 1.96 Å for Ni).

D. Antiferromagnetic insulators

We have found the following magnetically compensated insulators with double perovskite structure: (i) La₂CrMoO₆ is isoelectronic to the spin-compensated metal La₂CrWO₆ but introducing Mo instead of W opens up a gap at the Fermi level. It is the greater itineracy of the 5*d* element that leads to the metallic state that we analyzed above. (ii) In analogy with Sr₂CrOsO₆,⁸ we have tried with a bigger alkaline-earth cation. The compound Ba₂CrOsO₆ is found to be an antiferromagnetic insulator, with configuration *d*³-*d*³. As an isovalent compound, Sr₂CrRuO₆ is calculated to be an antiferromagnetic insulator.

VI. SUMMARY

In this paper we have suggested two very nearly spin-compensated half metals (or half semimetals): K_2MnRhO_6 and La_2CrWO_6 , both with a formally d^3-d^3 configuration. It is found that, within a double perovskite structure, a d^3-d^3 electronic configuration promotes the chances to obtain a

nearly spin-compensated half metal, due to several factors. (i) No competition between FM and AF interactions occurs as happens when there is partial occupation of e_g states in one of the metal cations. (ii) Having the t_{2g} shell completely filled leads to a very small orbital angular moment, which works even further in favor of a very small magnetic-moment solution. (iii) Having less than half-filled d shells in both metal ions helps to avoid imbalance in the total moment that occurs when spin-orbit coupling is important, specially for 4d and 5d elements. Having both cation d shells less than half filled means that the orbital angular momenta of both cations will also orient antiparallel to the spin, preventing the system from having a larger net magnetic moment. (iv) The large S=3/2 spins make our calculations more realistic; or-

dering of low-spin ions is often more affected by quantum fluctuations. Both materials that we have discussed at length are designed *ab initio*, no experimental information on them yet exists to the best of our knowledge.

ACKNOWLEDGMENTS

We have benefited from discussion on this topic with J. Chakhalian. This project was supported by the DOE under Grant No. DE-FG02-04ER46111 and through interactions with the DOE's Predictive Capability for strongly correlated systems team of the Computational Materials Science Network. V.P. acknowledges financial support from Xunta de Galicia through the Human Resources Program.

^{*}victor.pardo@usc.es

¹K. I. Kobayashi, T. Kimura, H. Sawada, K. Terakura, and Y. Tokura, Nature (London) **395**, 677 (1998).

²M. I. Katsnelson, V. Yu. Irkhin, L. Chioncel, A. I. Lichtenstein, and R. A. de Groot, Rev. Mod. Phys. **80**, 315 (2008).

³ J. Chakhalian, J. W. Freeland, G. Srajer, J. Strempfer, G. Khaliullin, J. C. Cezar, T. Charlton, R. Dalgliesh, C. Bernhard, G. Cristiani, H.-U. Habermeier, and B. Keimer, Nat. Phys. **2**, 244 (2006).

⁴H. van Leuken and R. A. de Groot, Phys. Rev. Lett. **74**, 1171 (1995).

⁵W. E. Pickett, Phys. Rev. Lett. **77**, 3185 (1996).

⁶W. E. Pickett and H. Eschrig, J. Phys.: Condens. Matter 19, 315203 (2007).

⁷K. W. Lee and W. E. Pickett, EPL **80**, 37008 (2007).

⁸K. W. Lee and W. E. Pickett, Phys. Rev. B 77, 115101 (2008).

⁹G. Jackeli and G. Khaliullin, Phys. Rev. Lett. **102**, 017205 (2009).

¹⁰W. E. Pickett, Phys. Rev. B **57**, 10613 (1998).

¹¹ J. H. Park, S. K. Kwon, and B. I. Min, Phys. Rev. B **65**, 174401 (2002).

¹² Y. K. Wang and G. Y. Guo, Phys. Rev. B 73, 064424 (2006).

¹³M. S. Park and B. I. Min, Phys. Rev. B **71**, 052405 (2005).

¹⁴M. Nakao, Phys. Rev. B **74**, 172404 (2006).

¹⁵I. Galanakis, K. Ozdogan, E. Sasioglu, and B. Aktas, Phys. Rev. B 75, 172405 (2007).

¹⁶I. Galanakis, P. H. Dederichs, and N. Papanikolaou, Phys. Rev. B 66, 134428 (2002).

¹⁷I. Galanakis, P. H. Dederichs, and N. Papanikolaou, Phys. Rev. B

^{66, 174429 (2002).}

¹⁸P. C. Canfield, Nat. Phys. **4**, 167 (2008).

¹⁹ V. I. Anisimov, J. Zaanen, and O. K. Andersen, Phys. Rev. B 44, 943 (1991).

²⁰ E. R. Ylvisaker, W. E. Pickett, and K. Koepernik, Phys. Rev. B 79, 035103 (2009).

²¹ J. B. Goodenough, *Magnetism and the Chemical Bond* (IEEE, New York, 2001).

²²P. Hohenberg and W. Kohn, Phys. Rev. **136**, B864 (1964).

²³K. Schwarz and P. Blaha, Comput. Mater. Sci. **28**, 259 (2003).

²⁴E. Sjöstedt, L. Nördstrom, and D. J. Singh, Solid State Commun. 114, 15 (2000).

²⁵ J. P. Perdew and Y. Wang, Phys. Rev. B **45**, 13244 (1992).

²⁶ A. I. Liechtenstein, V. I. Anisimov, and J. Zaanen, Phys. Rev. B 52, R5467 (1995).

²⁷J. P. Perdew, K. Burke, and M. Ernzerhof, Phys. Rev. Lett. 77, 3865 (1996).

²⁸R. E. Rudd and W. E. Pickett, Phys. Rev. B **57**, 557 (1998).

²⁹R. Macquart, S. J. Kim, W. R. Gemmill, J. K. Stalick, Y. Lee, T. Vogt, and H. S. C. zur Loye, Inorg. Chem. 44, 9676 (2005).

³⁰W. Song, E. Zhao, J. Meng, and Z. Wu, J. Chem. Phys. **130**, 114707 (2009).

³¹I. I. Mazin, D. I. Khomskii, R. Lengsdorf, J. A. Alonso, W. G. Marshall, R. M. Ibberson, A. Podlesnyak, M. J. Martinez-Lope, and M. M. Abd-Elmeguid, Phys. Rev. Lett. 98, 176406 (2007).

³²P. Ganguly, N. Y. Vasanthacharya, C. N. R. Rao, and P. P. Edwards, J. Solid State Chem. 54, 400 (1984).

³³G. J. Zhang, R. Liu, Y. Yang, and Y. Q. Yia, Phys. Status Solidi A **160**, 19 (1997).